

no other products present. The structures of azetidines **9a-c** were obvious from their ir, nmr, and mass spectra. That the geminal hydrogens were in the 3 position in **9a** and **b** was indicated by the low-field (τ 5.4) broad triplet for H_a . In **9c** H_a could not be resolved, but the mass spectra showed the expected α cleavage at m/e 43 ($CH_3CH=NH\cdot^+$) and no peak at m/e 30 ($CH_2=NH\cdot^+$).

The regioselectivity in the formation of **3** over **4** and **5** cannot be equated with conformational effects⁹ governing a concerted loss of N_2 with ring enlargement, since the conformers of **2** are expected to be of nearly equal stability and should produce a mixture of **3**, **4**, and **5**.

The driving force for ring expansion in preference to formation of a cyclopropylimine **5** (Ph migration) must be relief of ring strain in going from a three- to a four-membered ring. Apparently, the reaction involves selective migration of the electron richer CH_2 (or CR_2) rather than the CCl_2 ring residue to the electron-deficient nitrogen atom.

Rate data on the pyrolysis of cyclopropyl azides **2** (Table I) indicate that in general cis substituents (see **2b** and **2c**) accelerate the reaction. This implies that the rate-determining step cannot be loss of N_2 unless the reaction is concerted. A concerted ring expansion with loss of nitrogen is in fact suggested by the negative entropy of activation ($\Delta S^\ddagger = -18$ eu for **2c** \rightarrow **3c**) and is in accord with data on pyrolysis of simple alkyl azides.¹⁰

Acknowledgment. Support of this work by a grant (CA 04474) from the National Cancer Institute (Public Health Service) is gratefully acknowledged.

(9) Different migratory aptitudes in the photolysis of organic azides have been ascribed to ground-state conformational preferences; see R. M. Moriarty and R. C. Reardon, *Tetrahedron*, **26**, 1379 (1970).

(10) W. Lwowski, "Nitrenes," Interscience, New York, N. Y., 1970, Chapters 3 and 4.

(11) NIH Predoctoral Fellow, 1968-1971.

A. B. Levy,¹¹ A. Hassner*

Department of Chemistry, University of Colorado
Boulder, Colorado 80302

Received December 5, 1970

Effect of CH Scalar Coupling on ^{13}C Transverse Relaxation Times

Sir:

In view of the increasing interest in ^{13}C nmr spectroscopy, we wish to report some preliminary relaxation studies of ^{13}C -enriched compounds. We find that the

transverse relaxation times, T_{2C} ($\equiv R_{2C}^{-1}$), are usually shorter than the longitudinal relaxation times, T_{1C} ($\equiv R_{1C}^{-1}$). This result influences the sensitivity and resolution in both conventional and Fourier transform (FT) ^{13}C spectra and, in particular, affects the signal enhancement obtainable by the use of refocusing schemes such as driven equilibrium and spin-echo Fourier transform (DEFT and SEFT) techniques.¹⁻³

The T_{2C} measurements were performed at 15.08 MHz by the usual Meiboom-Gill⁴-modified Carr-Purcell⁵ technique on a high-resolution magnet with external fluorine lock. The π pulse width was 6 μ sec and the inhomogeneity over the sample gave a free-induction decay constant $T_2^* \sim 100$ msec. Degassed and sealed sample tubes were constricted below the liquid surface to minimize interchange between liquid and vapor. T_{1C} and T_{1H} values were measured by π - τ - $\pi/2$ pulse sequences, T_{1C} 's at 15.08 MHz, T_{1H} 's at 23.2 MHz. The T_1 and T_2 values were obtained by the usual least-squares fitting procedures.

Table I shows that T_{2C} is substantially shorter than

Table I. Relaxation Times^a for 60% ^{13}C -Enriched Compounds at 25°

| | T_{1C} , sec | T_{2C} , sec | T_{1H} , ^b sec |
|---------------------------------|-----------------|----------------|-----------------------------|
| $^{13}CH_3I$ | 13.4 | 3.9 | 10.9 ^c |
| $^{13}CH_3COOCD_3$ ^d | 19.2 | 6.1 | 12.5 |
| $^{13}CS_2$ | 48 ^e | 38 | |

^a The experimental error is estimated to be $\sim 5\%$. ^b T_{1H} for 100% enriched compounds will be less than these values. ^c T_{1H} for $^{12}CH_3I$ at 25° is reported [M. D. Zeidler, *Ber. Bunsenges. Phys. Chem.*, **69**, 659 (1965)] to be 12.5 sec. ^d Deuterium enrichment is 99%. ^e $T_{1C} = 30$ sec (15 MHz, 26-28°) was measured by rapid adiabatic passage; A. Olivson, E. Lippmaa, and J. Past, *Eesti NSV Tead. Akad. Toim.*, **16** 390 (1967).

T_{1C} ⁶ for methyl iodide and methyl- d_3 acetate- ^{13}C . We would like to point out that short T_{2C} 's occur when relaxing spin- $1/2$ nuclei, e.g., protons, are coupled to ^{13}C such that

$$A \equiv 2\pi J \gg R_{1H} \quad (1)$$

where $R_{1H} \equiv 1/T_{1H}$. Gutowsky, Vold, and Wells⁸ have pointed out the analogy between chemically exchanging, equally populated, uncoupled two-site systems and relaxing, scalar-coupled AX systems. They suggest that the two cases are related by the simple substitution of $\Omega \rightarrow A/2$ and $k \rightarrow R_{1X}$, where Ω is half the chemical shift between the two sites, k the exchange rate, and R_{1X} the spin-lattice relaxation rate of X.

Scalar contributions to R_{2A} for a general AX_n system originate from relaxation processes within the X manifold which preserve the spin state of A while changing the scalar coupling energy by a nonzero multiple of J . These processes randomly alter the precession frequency

(1) E. D. Becker, J. A. Ferretti, and T. C. Farrar, *J. Amer. Chem. Soc.*, **91**, 7784 (1969).

(2) A. Allerhand, *ibid.*, **92**, 4482 (1970).

(3) J. S. Waugh, *J. Mol. Spectrosc.*, **35**, 298 (1970).

(4) S. Meiboom and D. Gill, *Rev. Sci. Instrum.*, **29**, 688 (1958).

(5) H. Y. Carr and E. M. Purcell, *Phys. Rev.*, **94**, 630 (1954).

(6) We realize that T_1 is not rigorously defined for strongly coupled AX_n systems (see, e.g., ref 7, p 295) but we could detect no deviation from exponential behavior.

(7) A. Abragam, "The Principles of Nuclear Magnetism," Oxford University Press, London, 1961, p 311.

(8) H. S. Gutowsky, R. L. Vold, and E. J. Wells, *J. Chem. Phys.*, **43**, 4107 (1965).

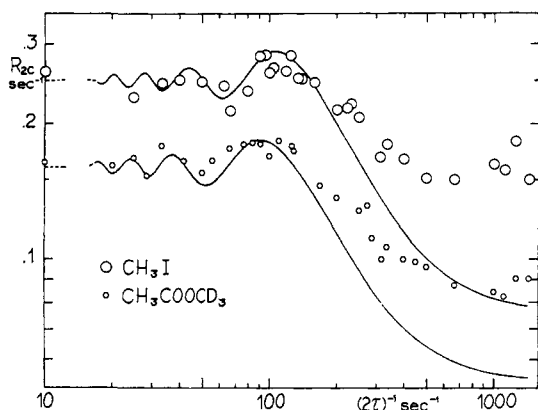


Figure 1. Apparent ^{13}C transverse relaxation rates *vs.* Carr-Purcell pulse repetition rate $(2\tau)^{-1}$ for 60% enriched $^{13}\text{CH}_3\text{I}$ and $^{13}\text{CH}_3\text{COOCD}_3$. The theoretical curves assume the model of a two-spin AX system, where $R_{2\text{C}}^0 = R_{1\text{C}}$ and \bar{R} is chosen to agree with the slow pulse limit. This model is based on $\Delta I_{\text{H}} = \pm 1$ transition only; however, inclusion of $\Delta I_{\text{H}} = \pm 2$ would improve the agreement in the region above $(2\tau)^{-1} = 100 \text{ sec}^{-1}$.

of A causing irreversible decay of the magnetization in the transverse plane. In particular, for $^{13}\text{CH}_3\text{I}$ and $^{13}\text{CH}_3\text{COOCD}_3$, whose $T_{1\text{H}}$'s are dominated by intramolecular dipolar effects, the scalar contribution to $R_{2\text{C}}$ is determined by the rate, \bar{R} , averaged over all states, of such H transitions. One might guess that $\bar{R} = 3R_{1\text{H}}$; however, $R_{1\text{H}}$ is based not only on the number of such transitions, but also on the ability of these to produce thermal equilibrium. Thus, for example, a $\Delta I_{\text{H}} = \pm 2$ transition is a more effective $T_{1\text{H}}$ process than $\Delta I_{\text{H}} = \pm 1$, whereas \bar{R} considers both equally. This implies that $\bar{R} < 3R_{1\text{H}}$ for a CH_3 system; to be more quantitative is beyond the scope of this communication.

The theory for coupled AX systems presented by Gutowsky, *et al.*,⁸ predicts a sigmoidal plot^{8,9} of $R_{2\text{C}}$ *vs.* $(2\tau)^{-1}$ for Carr-Purcell measurements with π pulse repetition rates $(2\tau)^{-1}$. Two limits are predicted. For rapid pulse rates

$$R_{2\text{C}} \rightarrow R_{2\text{C}}^0 \quad (2\tau)^{-1} \gg A \quad (2)$$

where $R_{2\text{C}}^0$ is the transverse relaxation rate in the absence of scalar coupling effects and is often very close to $R_{1\text{C}}$. For slow pulse rates

$$R_{2\text{C}} \rightarrow R_{2\text{C}}^0 + \bar{R} \quad (2\tau)^{-1} \ll A \quad (3)$$

and a damped oscillation is predicted^{8,9} for $(2\tau)^{-1} \lesssim J$.

In Figure 1, $R_{2\text{C}}$ for $^{13}\text{CH}_3\text{I}$ and $^{13}\text{CH}_3\text{COOCD}_3$ is plotted *vs.* $(2\tau)^{-1}$. The values of J_{CH} are 152 and 130 Hz, respectively, and there is good qualitative agreement between these curves and those predicted by theory.^{8,9} A maximum in $R_{2\text{C}}$ is predicted for $(2\tau)^{-1} = 2J/3$; we observe a maximum in the vicinity of this value. We also find a less rapid decrease in $R_{2\text{C}}$ at higher pulse rates, which is consistent with the occurrence of intramolecular $\Delta I_z = \pm 2$ transitions between proton spin states resulting in coupling energy changes of $\pm 2J$. Pulse repetition rates faster than those experimentally available to us [$(2\tau)^{-1} > 1000 \text{ sec}^{-1}$] are necessary to pulse these effects completely away. An unsatisfactory feature of this graph is that the values of $R_{2\text{C}}$ seem to plateau above $R_{1\text{C}}$ at the higher pulse rates. For

(9) A. Allerhand and H. S. Gutowsky, *J. Chem. Phys.*, **42**, 1587 (1965).

$^{13}\text{CH}_3\text{I}$ the scalar coupling to ^{127}I may result in $R_{2\text{C}}^0 > R_{1\text{C}}$; however, the higher limiting value of $R_{2\text{C}}^0$ observed for $^{13}\text{CH}_3\text{COOCD}_3$ cannot be explained this way. At these high pulse rates we may be exceeding the duty cycle of our transmitter; we plan to investigate compounds where the fast pulsing limit should occur well within our instrumental capabilities. In the slow pulsing limit our experimental values indicate, as expected, that $\bar{R} < 3R_{1\text{H}}$, when $R_{2\text{C}}^0 = R_{1\text{C}}$. Corrections required to account for scalar coupling effects from I in $^{13}\text{CH}_3\text{I}$ would decrease \bar{R} even further. The small ^{13}C -D coupling in $^{13}\text{CH}_3\text{COOCD}_3$ can in principle contribute to $R_{2\text{C}}$, but any effect has been pulsed away in these measurements [$(2\tau)^{-1} \geq 10 \text{ sec}^{-1}$].

As expected, $T_{2\text{C}}$ of $^{13}\text{CS}_2$ was found to be independent of the Carr-Purcell time. Experiments in this laboratory¹⁰ have shown that spin rotation is the dominant relaxation mechanism for $^{13}\text{CS}_2$ at 15 MHz, in agreement with data reported elsewhere.¹¹

In summary, it appears that the ultimate resolution in ^{13}C spectra will be of the order of the proton line widths, as long as eq 1 is satisfied for all coupled nuclei. Furthermore, the advantage predicted by the multiple-pulse, refocusing method^{1,2} over the normal FT technique will be greatly reduced, since the promised sensitivity enhancement³ was based on the assumption that $T_2 \approx T_1$.

The foregoing remarks relate to uncoupled spectra only. For pseudorandom proton noise decoupling the ^{13}C line widths should be a function of the H_2 rf level. At lower rf power, for which $A \gtrsim \tau_{\text{H}}^{-1}$, where τ_{H} is the effective lifetime of the proton spin states in the presence of H_2 , $R_{2\text{C}}$ should increase and reach a maximum as the rf power increases. For rf power sufficiently high that $A \ll \tau_{\text{H}}^{-1}$, $R_{2\text{C}}$ should then decrease toward $R_{2\text{C}}^0$. The refocusing techniques might therefore still be useful under conditions of strong, random proton decoupling.

(10) T. C. Farrar, S. J. Druck, R. R. Shoup, and E. D. Becker, to be published.

(11) In general, our I_2 measurements are reproducible to $\sim \pm 5\%$; however, systematic errors may increase for very fast pulse rates and very long T_2 's. The latter may account for the difference between $T_{1\text{C}}$ and $T_{2\text{C}}$ for $^{13}\text{CS}_2$.

R. R. Shoup*

National Institutes of Health
Bethesda, Maryland 20014

D. L. VanderHart

National Bureau of Standards
Washington D. C. 20234

Received November 30, 1970

Polarized Single-Crystal Absorption Spectrum of Cytosine Monohydrate

Sir:

In this communication we present polarized absorption spectra of single crystals of cytosine monohydrate. Callis and Simpson have recently reported a polarized reflectance study on cytosine monohydrate and 1-methylcytosine.¹ The results of these authors are in excellent agreement with our own, but our more sensitive direct absorption measurements using a microspectrophotometer permit additional and more quantitative

(1) P. R. Callis and W. T. Simpson, *J. Amer. Chem. Soc.*, **92**, 3593 (1970).